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2	57656	polyacrylamide	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:19
3	29523	"AMPS" or acrylamido adj4 (sulfonic adj acid)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:21
4	126938	isopropyl same alcohol or isopropanol	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:22
5	166860	peroxide	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:23
6	239868	acrylate or methacrylate	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:23
1	7	polyacrylamide adj prepolymer	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:25
7	902	((polyacrylamide adj prepolymer) or polyacrylamide) and ("AMPS" or acrylamido adj4 (sulfonic adj acid))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:25
8	190	(isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) and ("AMPS" or acrylamido adj4 (sulfonic adj acid)))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:25
9	84	peroxide and ((isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) and ("AMPS" or acrylamido adj4 (sulfonic adj acid))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:25
10	71	(acrylate or methacrylate) and (peroxide and ((isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) and ("AMPS" or acrylamido adj4 (sulfonic adj acid)))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:26
11	4315	((polyacrylamide adj prepolymer) or polyacrylamide) same (acrylate or methacrylate)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:26
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13	231	((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:27
14	46	(isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid)))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:27
15	115	(acrylate or methacrylate) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid)))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:27

16	27	((acrylate or methacrylate) and ((isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid)))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:28
17	13	peroxide and ((acrylate or methacrylate) and ((isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid)))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:28
18	114	graft adj (initiator or catalyst)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
19	71	peroxide and (graft adj (initiator or catalyst))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
20	57656	(polyacrylamide adj prepolymer) or polyacrylamide	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
21	0	((polyacrylamide adj prepolymer) or polyacrylamide) and (peroxide and (graft adj (initiator or catalyst)))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:31
22	0	Freetex\$ same "695"	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:31
23	975	polyacrylamide adj polymer	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:32
24	902	((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:33
25	2590	hydroxy adj ethyl adj (acrylate or methacrylate)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:34
26	11	(((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid))) and (hydroxy adj ethyl adj (acrylate or methacrylate))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:34

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L29: (17) 5 and 28
L31: (529) 6 and 24
L32: (138) 4 and 31
L33: (71) 5 and 32
L30: (13) 6 and 29
L34: (1113259) coating or coating adj cor
L35: (19) 33 and 34

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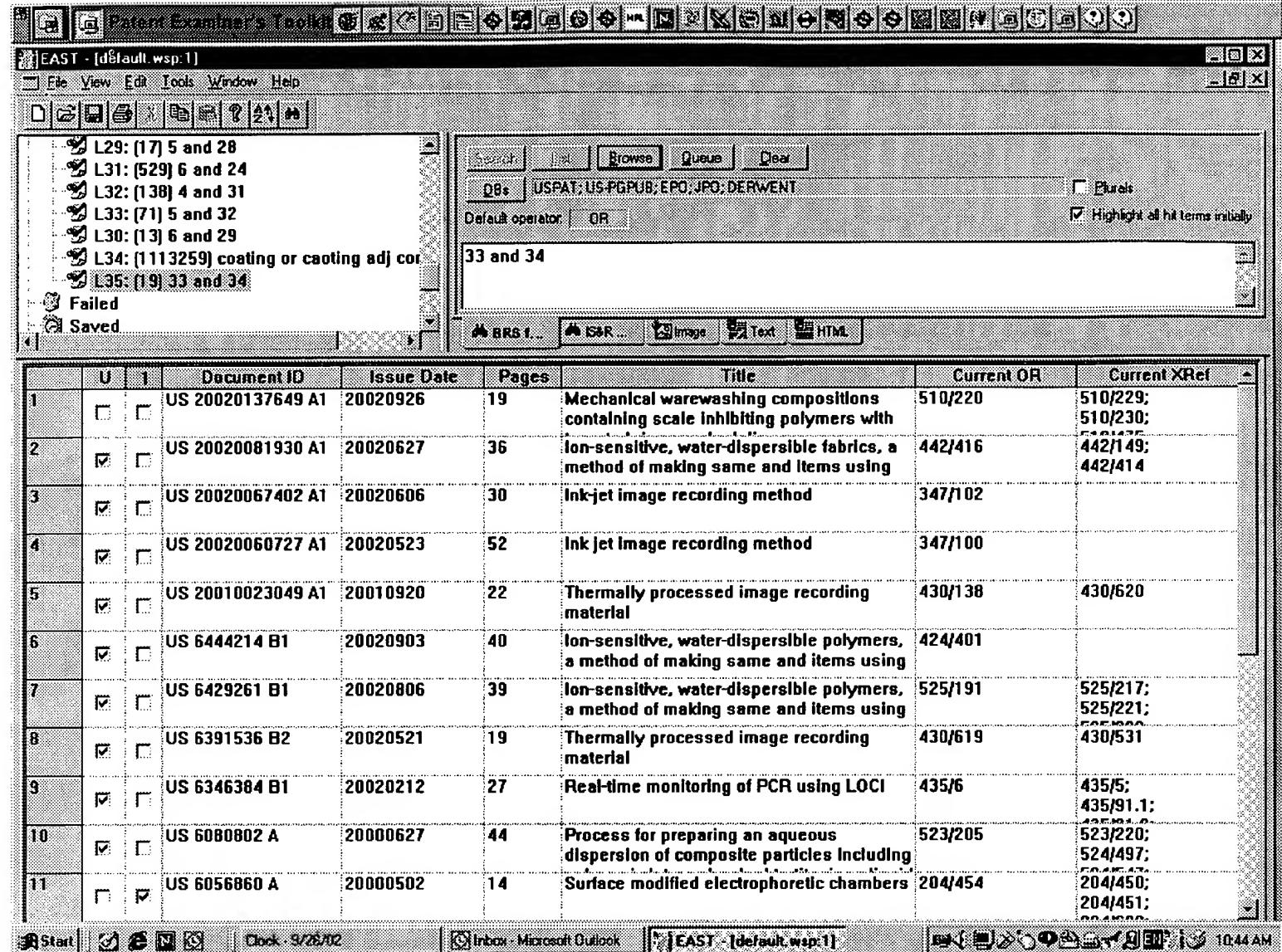
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11	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 6056860 A	20000502	14	Surface modified electrophoretic chambers	204/454	204/450; 204/451;
12	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 5387318 A	19950207	7	Water soluble graft copolymers for laser print deinking loop clarification	162/5	162/189; 162/190;
13	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 5338490 A	19940816	15	Two-phase composites of ionically-conductive pressure-sensitive	252/500	424/448; 524/394;
14	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 5277823 A	19940111	15	Silica scale inhibition	210/696	127/61; 166/244.1;
15	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 5108622 A	19920428	9	Block copolymers for sludge dewatering and a polymeric initiator for their	210/734	210/732; 210/733
16	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4879361 A	19891107	13	Preparation of bead polymers from water-soluble, ethylenically unsaturated	526/201	526/213
17	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 4708997 A	19871124	7	Suspending agent for the suspension polymerization of water-soluble monomers	526/207	524/801; 526/206
18	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4518687 A	19850521	16	Silver halide color photographic light-sensitive material	430/548	430/381; 430/552;
19	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 4194998 A	19800325	6	Highly absorbent polyhydroxy polymer graft copolymers without saponification	527/314	106/206.1; 210/689;

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 L25: (2590) hydroxy adj ethyl adj (acryla
 L26: (11) 24 and 25
 L27: (231) [1 or 2 or 23] same 3
 L28: (46) 4 and 27
 L31: (529) 6 and 24
 L32: (138) 4 and 31
 L33: (711) 5 and 32

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 DBs USPAT; US-PCPUB; EPO; JPO; DEPATENT
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24 and 25

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2	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 20010038831 A1	20011108	31	Super-absorbent hydrogel foams	424/78.31	424/78.35; 424/78.37;
3	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 6271278 B1	20010807	29	Hydrogel composites and superporous hydrogel composites having fast swelling.	521/150	521/102; 521/109.1;
4	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 6020209 A	20000201	6	Polymers for drilling and reservoir fluids and their use	507/120	507/118; 507/119
5	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 5882677 A	19990316	9	Iontophoretic patch with hydrogel reservoir	424/449	424/443; 424/444;
6	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 5663123 A	19970902	36	Polymeric earth support fluid compositions and method for their use	507/225	507/118; 507/120;
7	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 5593809 A	19970114	13	Peel apart diffusion transfer compound film unit with crosslinkable layer and borate	430/213	430/215; 430/216;
8	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 4589964 A	19860520	14	Process for graft copolymerization of a pre-formed substrate	522/85	427/302; 427/322;
9	<input type="checkbox"/>	<input checked="" type="checkbox"/>	US 4459318 A	19840710	6	Method for forming a self-lubricating fill tube	427/551	138/109; 138/145;
10	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4417992 A	19831129	7	Dust control	252/88.1	299/12; 404/76;
11	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4311573 A	19820119	7	Process for graft copolymerization of a pre-formed substrate	522/129	522/134; 525/245;

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Patent Examiner's Toolkit

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L35: (19) 33 and 34
L29: (17) 5 and 28
L17: (13) 5 and 16
L10: (71) 6 and 9

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DBs: USPAT; US-PCGPUB; EPO; JPO; DERWENT
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6 and 9

BRS 1... ISR... Image Text HTML

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7	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 6444214 B1	20020903	40	Ion-sensitive, water-dispersible polymers, a method of making same and items using	424/401	
8	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 6429261 B1	20020806	39	Ion-sensitive, water-dispersible polymers, a method of making same and items using	525/191	525/217; 525/221;
9	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 6395853 B1	20020528		Water-soluble copolymers and their use for exploration and production of petroleum	526/307.2	526/274; 526/287;
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14	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 6177478 B1	20010123	10	Method for reducing oxalate	514/789	

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5	166860	peroxide	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:23
6	239868	acrylate or methacrylate	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:23
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13	231	((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:27
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15	115	(acrylate or methacrylate) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid)))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:27
16	27	(acrylate or methacrylate) and ((isopropyl same alcohol or isopropanol) and (((polyacrylamide adj prepolymer) or polyacrylamide) same ("AMPS" or acrylamido adj4 (sulfonic adj acid))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:28

18	114	graft adj (initiator or catalyst)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
19	71	peroxide and (graft adj (initiator or catalyst))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
20	57656	(polyacrylamide adj prepolymer) or polyacrylamide	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:29
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22	0	Freetex\$ same "695"	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:31
23	975	polyacrylamide adj polymer	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:32
24	902	((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:33
25	2590	hydroxy adj ethyl adj (acrylate or methacrylate)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:34
26	11	((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid)) and (hydroxy adj ethyl adj (acrylate or methacrylate))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:36
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33	71	peroxide and ((isopropyl same alcohol or isopropanol) and ((acrylate or methacrylate) and (((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid)))))	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:39
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34	1113259	coating or caoting adj composition	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:40
35	19	(peroxide and ((isopropyl same alcohol or isopropanol) and ((acrylate or methacrylate) and (((polyacrylamide adj prepolymer) or polyacrylamide or (polyacrylamide adj polymer)) and ("AMPS" or acrylamido adj4 (sulfonic adj acid)))))) and (coating or caoting adj composition)	USPAT; US-PGPUB; EPO; JPO; DERWENT	2002/09/28 10:40
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63		<input type="checkbox"/>	US 4637418 A	19870120	5	Fluid friction reduction	137/13	406/47; 406/49;
64	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4599390 A	19860708	23	High molecular weight water-soluble polymers and flocculation method using	526/240	210/734; 210/907;
65	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4569768 A	19860211	7	Flocculation of suspended solids from aqueous media	210/727	209/5; 210/734
66	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4518687 A	19850521	16	Silver halide color photographic light-sensitive material	430/548	430/381; 430/552;
67		<input checked="" type="checkbox"/>	US 4507440 A	19850326	16	Cross-linkable and cross linked macromolecular compositions wherein	525/218	166/307; 166/308;
68	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4499232 A	19850212	14	Water soluble, crosslinkable polymer compositions, their preparation and use	524/548	166/307; 166/308;
69	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4432881 A	19840221	14	Water-dispersible hydrophobic thickening agent	507/121	166/275; 507/118;
70	<input checked="" type="checkbox"/>	<input type="checkbox"/>	US 4339371 A	19820713	7	High concentration water-soluble polymers in water-in-oil emulsions	524/310	524/458; 524/460;
71	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>	US 4194998 A	19800325	6	Highly absorbent polyhydroxy polymer graft copolymers without saponification	527/314	106/206.1; 210/689;



US005387318A

United States Patent [19]

Liao et al.

[11] Patent Number: 5,387,318

[45] Date of Patent: * Feb. 7, 1995

[54] **WATER SOLUBLE GRAFT COPOLYMERS FOR LASER PRINT DEINKING LOOP CLARIFICATION**

[75] Inventors: Wen P. Liao, Warminster, Pa.; John C. Harrington, Jacksonville, Fla.; Walter J. Burgess, Phoenixville; Fu Chen, Newtown, both of Pa.

[73] Assignee: **Betz Laboratories, Inc., Trevose, Pa.**

[*] Notice: The portion of the term of this patent subsequent to May 18, 2010 has been disclaimed.

[21] Appl. No.: 169,819

[22] Filed: Dec. 17, 1993

Related U.S. Application Data

[60] Continuation-in-part of Ser. No. 773, Jan. 5, 1993, Pat. No. 5,298,566, which is a division of Ser. No. 691,206, Apr. 25, 1991, Pat. No. 5,211,854.

[51] Int. Cl.⁶ C02F 1/56

[52] U.S. Cl. 162/5; 162/72; 162/189; 162/190; 210/734

[58] Field of Search 210/734; 525/296; 162/5, 72, 189, 190

References Cited**U.S. PATENT DOCUMENTS**

5,013,456	5/1991	St. John	210/734
5,211,854	5/1993	Liao et al.	210/734
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5,269,942	12/1993	Harrington	210/727

OTHER PUBLICATIONS

Pulp & Paper Chemistry & Chemical Technology, 3d ed., vol. III, 1981, pp. 1593-1607.

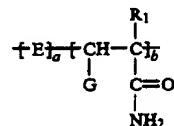
Betz Handbook of Industrial Water Conditioning, 9th ed., 1991, pp. 22-30.

Primary Examiner—Vasu S. Jagannathan

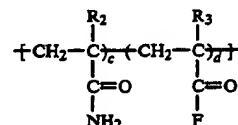
Attorney, Agent, or Firm—Alexander D. Ricci; Philip H. Von Neida; Gregory M. Hill

[57] ABSTRACT

A method for clarifying the laser print deinking loop water in a papermaking process by using a water soluble graft copolymer having the structure:



wherein E is the repeat unit obtained after polymerization of an α,β ethylenically unsaturated compound, the molar percentage of a:b is from about 95:5 to 5:95, with the proviso that the sum of a and b equals 100%; G comprises the structure:



wherein d is a cationic monomer, R₁, R₂ and R₃ are the same or different and are hydrogen or a lower alkyl group having C₁ to C₃, F is the salt of an ammonium cation and the molar percentage of c:d is from 95:5 to 5:95 with the proviso that the sum of c and d equals 100%.

13 Claims, 2 Drawing Sheets

FIG. 1

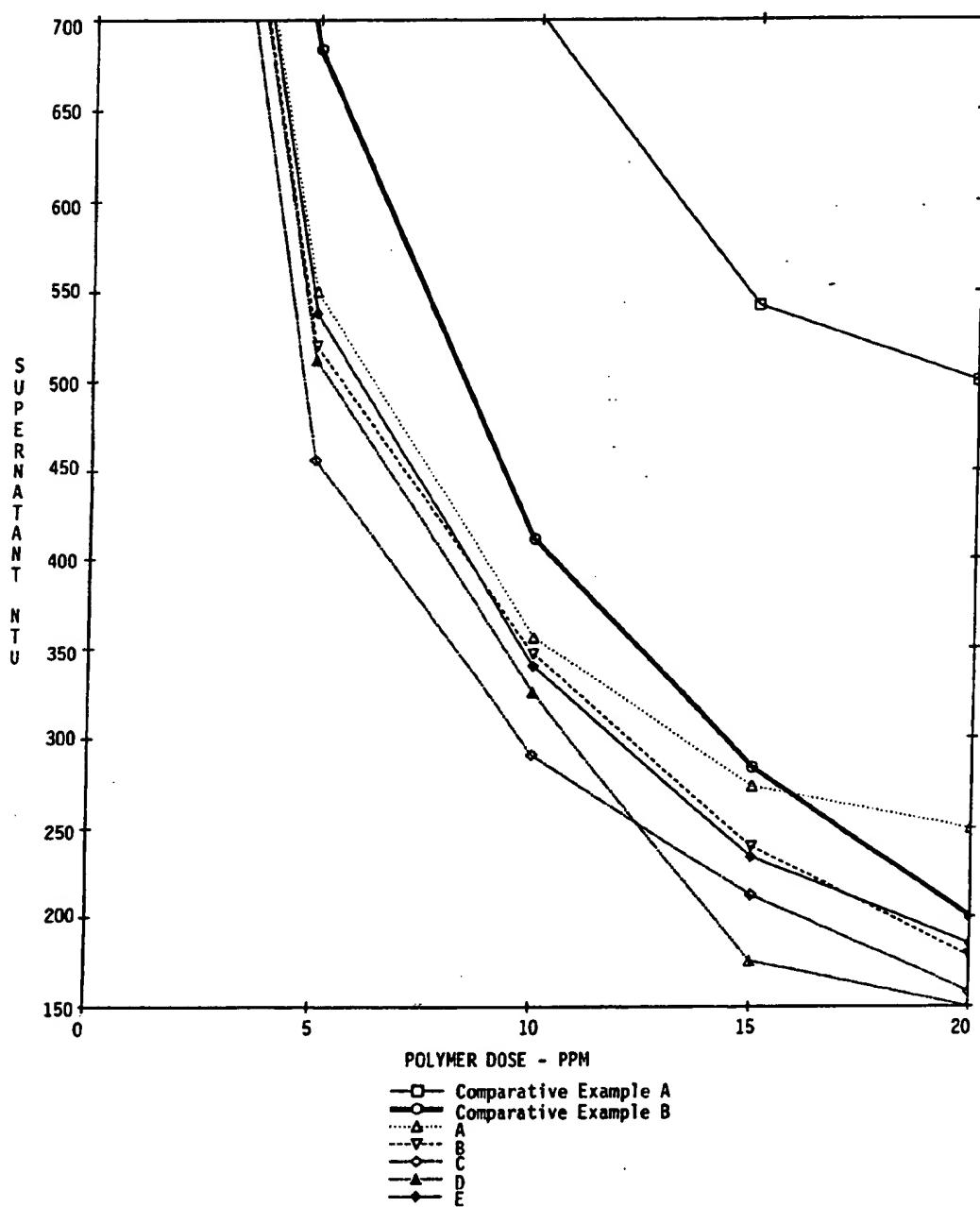
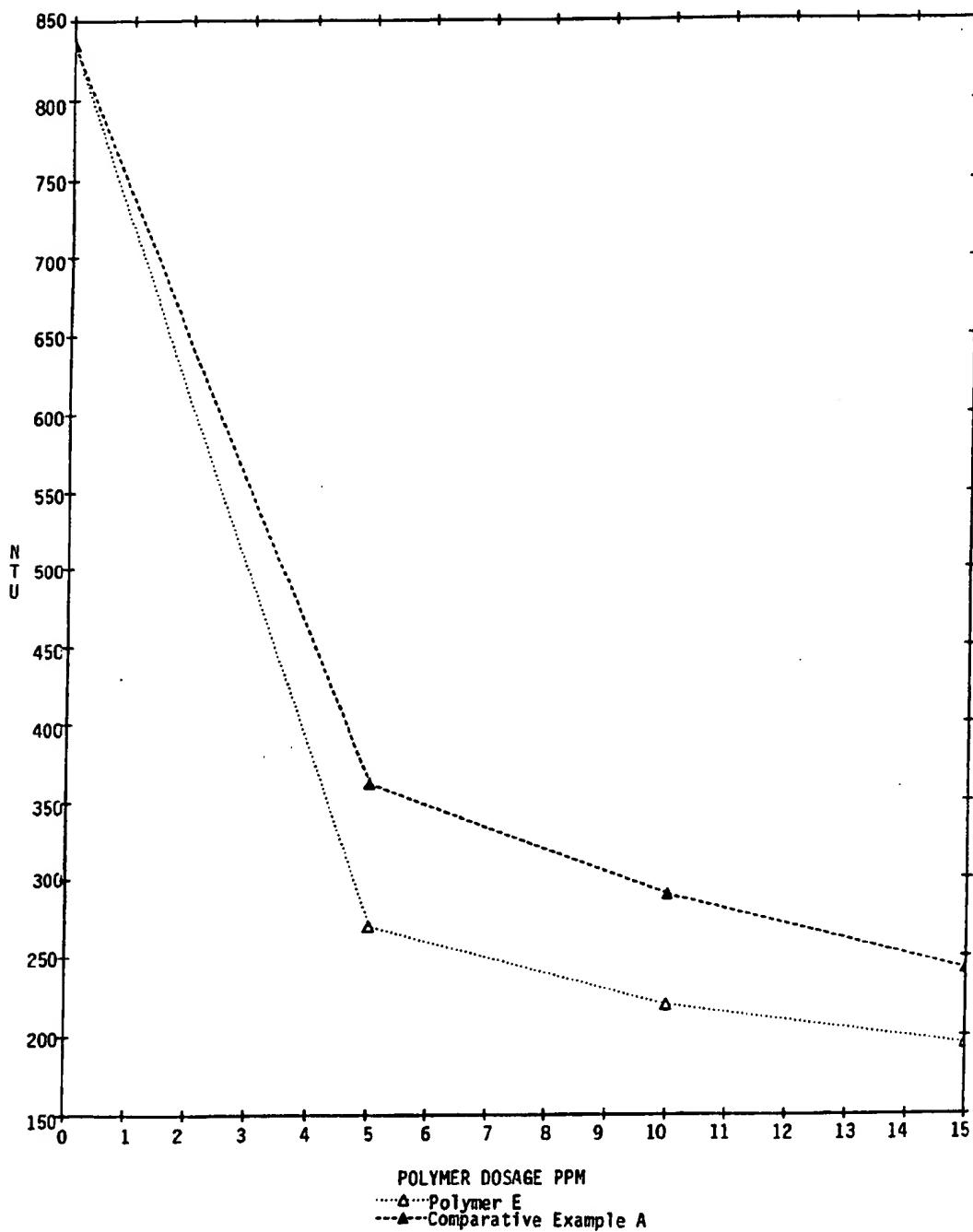


FIG. 2



**WATER SOLUBLE GRAFT COPOLYMERS FOR
LASER PRINT DEINKING LOOP
CLARIFICATION**

This is a continuation-in-part of application Ser. No. 08/000,773 filed Jan. 5, 1993 now U.S. Pat. No. 5,298,566, which is a divisional of application Ser. No. 07/691,206 filed Apr. 25, 1991 now U.S. Pat. No. 5,211,854.

FIELD OF THE INVENTION

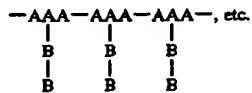
The present invention pertains to novel water soluble graft copolymers which are useful for water treatment, such as sludge dewatering and water clarification. More particularly, it relates to the use of a novel graft copolymer for the clarification of water in the deinking loop of a paper making process using recycled laser print paper.

BACKGROUND OF THE INVENTION

There is an increasing usage of water soluble polymers and copolymers in wastewater treatment industries. These compounds have shown desirable utility for the purpose of dewatering sludge and clarifying contaminated water.

The efficacies of the polymers or copolymers used will vary depending upon the type of monomers chosen to form the polymer or copolymer, the molecular weight of the synthesized molecule and, in the case of a copolymer, the placement of the selected monomers on the backbone of the copolymer. It is the latter characteristic that is the focus of the present invention.

Polymers with long sequences of two monomers can be categorized as block copolymers or graft copolymers. In graft copolymers sequences of one monomer are "grafted" onto a "backbone" of the second monomer type.



Graft copolymers have unique and highly desirable properties as compared to random copolymers or the blend of two homopolymers. Therefore, there is a great interest in preparing them. Few techniques described in the literature satisfy the need.

Furthermore, with ever increasing usages of water soluble polymers and copolymers in industries such as wastewater treatment, cooling, boiler and deposit control, coating, textile, mining, detergency, cosmetics, and papermaking, etc., there is an urgent need to synthesize novel water soluble graft copolymers for this broad range of applications.

More specifically, the use of recycled fibers is becoming an important aspect of papermaking for economic and environmental considerations. The preliminary manufacturing steps in the use of recycled fibers for papermaking consists of repulping the paper sources, then removing the printing inks from the fibers. A typical deinking process utilizes a combination of chemical and mechanical techniques in several stages. Large amounts of water are used in the washing or flotation stages, wherein chemically treated ink particles and other contaminants are physically removed from the fibrous slurry. The wastewater, or effluent, from these stages is typically recycled back into the mill for reuse

in the deinking process. As the recycled effluent contains dispersed inks, fiber fines and inorganic fillers, these contaminants must be removed to provide a clean water source for the deinking process and to prevent the dispersed inks from being reintroduced into the fibers. The effluent may also be discharged from the mill; thus, suspended solids must be removed from the wastewater to meet environmental regulations.

Clarification chemicals are typically utilized in conjunction with mechanical clarifiers for the removal of solids from the effluent. Clarification generally refers to the removal of material by coagulation, and/or flocculation, then sedimentation or flotation. See the Betz Handbook of Industrial Water Conditioning 9th Edition, 1991, Betz Laboratories, Inc., Trevose, Pa. pages 23 through 30.

Conventional polyacrylamide copolymers have been used in this application. However, there still exists a need to provide a novel polymer in a more effective and economic treatment process. This objective is achieved by the present invention. The novel graft copolymers exhibit the desired efficacy for laser deink clarification applications.

BRIEF DESCRIPTION OF THE DRAWING

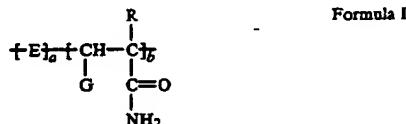
FIGS. 1 and 2 are graphs demonstrating water clarification (supernatant NTU) versus polymer dosage for the inventive graft copolymers and comparative linear polymers.

**DETAILED DESCRIPTION OF THE
INVENTION**

The present invention pertains to the use of novel water soluble graft copolymers as laser print deinking loop clarifiers.

Specifically, the graft polymers in the invention contain polymeric segments obtained from the polymerization of acrylamide and cationic monomers which are attached or "grafted" to another polymer chain which is comprised of the repeating units of one or more monomers. The resulting graft copolymers are soluble in an aqueous medium.

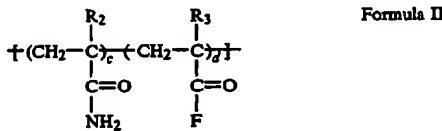
The graft copolymer of the invention has the general structure:



wherein E in the above formula (Formula I) is the repeat unit obtained after polymerization of an α,β ethylenically unsaturated compound, preferably carboxylic acid, amide form thereof, alkyl (C1-C8) ester or hydroxylated alkyl (C1-C8) ester of such carboxylic acid. Compounds encompassed by E include the repeat unit obtained after polymerization of acrylamide, methacrylamide, acrylic acid, methacrylic acid, maleic acid or anhydride, styrene sulfonic acid, 2-acrylamido-2-methylpropyl sulfonic acid, itaconic acid, and the like. Ester derivatives of the above mentioned acids such as 2-hydroxypropyl acrylate, methyl methacrylate, and 2-ethylhexyl acrylate, are also within the purview of the invention.

The molar percentage of a:b is from about 95:5 to 5:95, with the proviso that the sum of a and b equals 100%.

G in the above formula (Formula I) is a polymeric segment comprising repeat units having the structure:



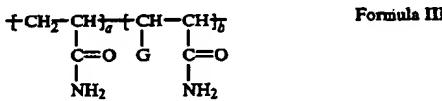
wherein R₁, R₂ and R₃ in Formulae I and II are the same or different and are hydrogen or a lower alkyl group having C₁ to C₃. F in the above formula is a salt of an ammonium cation, such as NHR₃N⁺R(4,5,6)M⁻ or OR₃N⁺R(4,5,6)M⁻, wherein R₃ is a C₁ to C₄ linear or branched alkylene group, and R₄, R₅ and R₆ can be selected from the group consisting of hydrogen, C₁ to C₄ linear or branched alkyl, C₅ to C₈ cycloalkyl, aromatic or alkylaromatic group; and M is an anion, such as chloride, bromide, or methyl or hydrogen sulfate. Typical cationic monomers are 2-acryloyloxyethyltrimethylammonium chloride (AETAC), 3-(methacrylamidopropyltrimethylammonium chloride (MAP-TAC or APTAC), 2-methacryloyloxyethyltrimethylammonium chloride (METAC) and diallyldimethylammonium chloride (DADMAC), etc.

It is understood that more than one kind of cationic monomer may be present in Formula II.

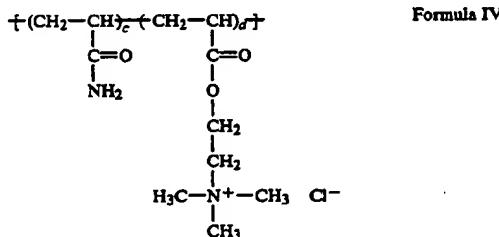
The molar percentage c:d in Formula II may vary from 95:5 to 5:95, with the proviso, however, the sum of c and d equals 100%.

There is no limit to the kind and mole percent of the monomers chosen so long as the total adds up to 100 mole % and the resulting copolymers are water soluble.

At present, the preferred water soluble graft copolymer for use as a laser print deinking loop clarifier is:



The molar percentage of a:b is from about 95:5 to 5:95, with the proviso that the sum of a and b equals 100%. G in Formula III is:



The cationic monomer is 2-acryloyloxyethyltrimethylammonium chloride (AETAC). The molar percentage c:d in the polymer segment G (Formula IV) is the ratio of Acrylamide:AETAC. It may fall within the range between 95:5 and 5:95. The sum of c and d must add up to 100%.

The number average molecular weight (M_n) of the polymeric segment G is not critical and may fall within

the range of 1,000 to 1,000,000. Preferably, the number average molecular weight will be within the range of 5,000 to 500,000, with the range of about 10,000 to about 200,000 being even more desirable. The key criterion is that the resulting graft copolymer be water soluble.

The graft copolymer is prepared via a two-step polymerization process. First, a macromonomer comprised of acrylamide and AETAC is prepared by a water-in-oil inverse emulsion polymerization method using peroxide as an initiator. Such processes have been disclosed in U.S. Pat. Nos. 3,284,393, Re. 28,474 and Re. 28,576, herein incorporated by reference. The initiator may be selected from peroxides, persulfates, bromates, and azo-type initiators such as 2,2'-azobis-(2-amidinopropane) dihydrochloride, 2,2'-azobis-(2,4-dimethylvaleronitrile). Copper (II) sulfate is added in the process as an oxidative chain transfer agent to generate a terminal unsaturated double bond in the polymer chain. It is conceivable that transition metal ions other than copper, such as iron, cobalt, and nickel etc., may be used in the invention.

Ethylenediaminetetraacetic acid or diethylenetriamine pentaacetic acid and their salts or their amino analogue are used as chelating agents to chelate or to form complexes with copper prior to the second polymerization step.

The resulting macromonomer is then copolymerized with acrylamide or other monomers to form graft copolymers by a similar water-in-oil inverse emulsion technique.

Branching agents such as polyethyleneglycol dimethacrylate, N,N'-methylenebis(methacrylamide), N-vinyl acrylamide, allyl glycidyl ether, glycidyl acrylate and the like may also be added, providing the resulting graft copolymer is water soluble. Any of the well known chain transfer agents familiar to those who skilled in the art may be used to control the molecular weight. Those include, but are not limited to, lower alkyl alcohols such as isopropanol, amines, mercaptans, phosphites, thioacids, formate, allyl alcohol and the like.

Conventional initiators such as peroxide, persulfate, along with sulfite/bisulfite and azo compounds may be used depend on the system chosen.

High HLB inverting surfactants such as those described in U.S. Pat. No. Re. 28,474 are then added to the emulsion to convert the resulting emulsion to a "self-inverting" emulsion. Using the procedure described herein, a unique graft copolymer in emulsion form is obtained.

The resulting copolymer may also be further isolated by precipitating it in an organic solvent such as acetone and dried to a powder form. The powder can be easily dissolved in an aqueous medium for use in the desired applications.

It is to be understood that the aforementioned polymerization methods do not in any way limit the synthesis of copolymers according to this invention.

The resulting emulsion disperses and dissolves rapidly into an aqueous solution upon addition to water. Within minutes, a maximum solution viscosity is obtained. The emulsion dissolves well even in water containing a high level of hardness and it also retains most of its solution viscosity in brine water.

The structure of the graft copolymer is substantiated by a conventional solution viscosity study and C¹³ NMR spectra copy. The molecular weight of the re-

sulting graft copolymer is not critical, as long as the polymer is soluble in water. The molecular weight may vary over a wide range, e.g., 10,000-30,000,000 and may be selected depending upon the desired application.

The graft copolymer is added to the influent flow prior to the clarifier. It is added in an amount of from about 0.5 to 100 ppm of polymer per total influent volume. Preferably 2 to 25 ppm of polymer per total influent volume is used.

Experimental

Properties of five water soluble graft copolymers prepared according to the procedure described above are shown in Table I. The copolymers contain an overall amount of 20 mole % AETAC and 80 mole % acrylamide.

TABLE I

Physical Properties of the Graft Copolymers		
Polymer	Solids %	UL* Viscosity (cps)
A	39.6	9.0
B	35.8	11.4
C	38.5	12.2
D	41.4	18.1
E	33.2	25.1

*UL viscosity: 0.3% solids of polymer dissolved in 4% NaCl solution, as measured with an UL adapter in a Brookfield Viscometer.

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Performance Test

EXAMPLE 1

In the following test, the performance of the water soluble graft copolymers described in this invention is demonstrated. The test substrate is a deinking process water containing 100% recycled laser print fiber from a Midwest paper mill. The substrate had the following properties: pH, 7.0-7.5, solids, 0.126% and a total turbidity of 1300-1400 NTU.

TEST PROCEDURE—DEINK LOOP CLARIFICATION

1. 250 milliliters (ml) of stock at 25°C is measured in a graduated cylinder and poured into a 400 ml glass beaker. The beaker contains a Teflon coated magnetic stirring bar, and is centered on a magnetic stirring plate. The stir plates have been previously calibrated to provide approximately equivalent shear mixing speeds at "high" and "low" speeds.

2. The beakers are turned on to high speed; once the samples have equilibrated, the test level of coagulant is introduced into the center of the vortex with a previously filled syringe. The polymer is allowed to mix for a predetermined time consistent with the individual mill's clarifier design, typically 10 to 60 seconds.

3. After the polymer mix time, the speed of the mixer is reduced to "low" speed for a time period consistent with the actual clarifier, typically 30 to 60 seconds. After the low speed mixing time is completed, the mixers are turned off, and the flocculated particles are allowed to settle. The settling volumes and times are recorded.

4. Supernatant is then removed from the beaker, and the turbidity is recorded on a turbidimeter for each polymer and polymer dosage level.

The results are shown in Table II and FIG. 1. Comparative polymers A and B are commercially available linear copolymers containing 20 mole % and 40 mole % of AETAC, respectively (the remainder, acrylamide).

5 The above data demonstrate that the graft copolymers in this invention are more effective in water clarification than the comparative linear polymers.

TABLE II

10	Polymer	Clarification Test	
		Dosage (ppm)	Supernatant (NTU)
20	A	5	550
		10	356
		15	273
		20	250
25	B	5	520
		10	347
		15	240
		20	178
30	C	5	456
		10	291
		15	212
		20	158
35	D	5	512
		10	325
		15	175
		20	150
40	E	5	538
		10	340
		15	234
		20	185
45	Comparative Polymer A	5	860
		10	702
		15	542
		20	500
50	Comparative Polymer B	5	684
		10	411
		15	284
		20	200

EXAMPLE 2

Where the previous example tested deinking process water that contained 100% recycled laser print, many paper mills will only utilize a portion of laser print in the total fiber composition. In the following example, process water from a Northwest paper mill was tested. It contained approximately 20% to 40 recycled laser print fiber with the remainder being made up of nonimpact print fiber. The substrate exhibited a pH of 6.6 and had a solids content of 0.052%. The same test protocol as in Example 1 was followed and the polymers used are as defined previously. Results are shown in Table III and FIG. 2.

TABLE III

55	Polymer	Clarification Test	
		Dosage (ppm)	Supernatant (NTU)
55	Blank	0	835
		5	270
		10	220
		15	195
60	Comparative Example A	5	362
		10	290
		15	242

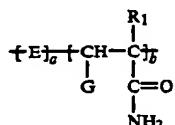
The foregoing tests demonstrate that the graft configuration of the subject polymer is more efficient at clarifying process water containing laser print fiber than is the linear configuration of the same molecule.

While this invention has been described with respect to particular embodiments thereof, it is apparent that

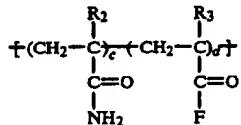
numerous other forms and modifications of this invention will be obvious to those skilled in the art. The appended claims and this invention generally should be construed to cover all such obvious forms and modifications which are within the true spirit and scope of the present invention.

We claim:

1. A method of clarifying laser print drinking loop water of a papermaking process comprising adding to the clarifier influent from about 0.5 ppm to about 500 ppm of a water soluble graft copolymer having the structure:



wherein E is the repeat unit obtained after polymerization of an α,β ethylenically unsaturated compound, the molar percentage of a:b is from about 95:5 to 5:95, with the proviso that the sum of a and b equals 100%; G comprises the structure:



wherein d is a cationic monomer, R₁, R₂ and R₃ are the same or different and are hydrogen or a lower alkyl group having C₁ to C₃, F is the salt of an ammonium cation and the molar percentage of c:d is from 95:5 to 5:95 with the proviso that the sum of c and d equals 100%.

2. The method of claim 1 wherein the α,β ethylenically unsaturated compound is selected from the group consisting of an unsaturated carboxylic acid, the amide form thereof, the alkyl (C₁-C₈) ester thereof and the hydroxylated alkyl (C₁-C₈) thereof, styrene sulfonic acid and 2-acrylamido-2-methyl propyl sulfonic acid.

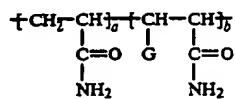
3. The method of claim 2 wherein the α,β ethylenically unsaturated compound is selected from the group consisting of acrylamide, methacrylamide, acrylic acid, methacrylic acid, maleic acid, maleic anhydride, itaconic acid, 2-hydroxypropyl acrylate, methyl methacrylate and 2-ethylhexyl acrylate.

4. The method of claim 1 wherein F is selected from the group consisting of NHR₃N⁺(R₄, R₅, R₆)M⁻ and OR₃N⁺(R₄, R₅, R₆)M⁻, wherein R₃ is a C₁ to C₄ linear or branched alkylene group, R₄, R₅, and R₆ are selected from the group consisting of hydrogen, C₁ to C₄ linear

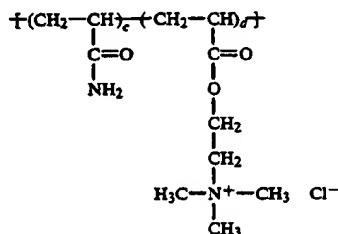
or branched alkyl, C₅ to C₈ cycloalkyl, aromatic or alkylaromatic group, and M⁻ is an anion selected from the group consisting of chloride, bromide, methyl sulfate and hydrogen sulfate.

5. The method of claim 4 wherein the cationic monomer is selected from the group consisting of 2-acryloyloxyethyltrimethylammonium chloride, 3-methacrylamidopropyltrimethyl ammonium chloride, 2-methacryloyloxyethyltrimethylammonium chloride and diallyldimethylammonium chloride.

6. The method of claim 1 wherein the graft copolymer has the structure:



20 wherein the molar percentage a:b is from about 95:5 to 5:95, with the proviso that the sum of a and b equals 100%, and G has the structure:



35 7. The method of claim 1 wherein the number average molecular weight of G is from about 1,000 to about 1,000,000.

8. The method of claim 7 wherein the number average molecular weight of G is from about 5,000 to about 500,000.

40 9. The method of claim 8 wherein the number average molecular weight of G is from about 10,000 to about 200,000.

10. The method of claim 1 wherein the graft copolymer has a number average molecular weight of from about 10,000 to 30,000,000.

45 11. The method of claim 10 wherein the graft copolymer has a number average molecular weight of from about 1,000,000 to 30,000,000.

50 12. The method of claim 1 wherein the amount of graft copolymer added is from about 2 to 25 ppm per total influent volume.

13. The method of claim 1 wherein the graft copolymer is added to the influent flow prior to the clarifier.

* * * * *